



Invited Review Article

The many types of carbonic anhydrases in photosynthetic organisms

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ABSTRACT

Carbonic anhydrases (CAs) are enzymes that catalyze the interconversion of CO₂ and HCO₃⁻. In nature, there are multiple families of CA, designated with the Greek letters α through θ. CAs are ubiquitous in plants, algae and photosynthetic bacteria, often playing essential roles in the CO₂ concentrating mechanisms (CCMs) which enhance the delivery of CO₂ to Rubisco. As algal CCMs become better characterized, it is clear that different types of CAs are playing the same role in different algae. For example, an α-CA catalyzes the conversion of accumulated HCO₃⁻ to CO₂ in the green alga *Chlamydomonas reinhardtii*, while a θ-CA performs the same function in the diatom *Phaeodactylum tricornutum*. In this review we argue that, in addition to its role of delivering CO₂ for photosynthesis, other metabolic roles of CA have likely changed as the Earth's atmospheric CO₂ level decreased. Since the algal and plant lineages diverged well before the decrease in atmospheric CO₂, it is likely that plant, algae and photosynthetic bacteria all adapted independently to the drop in atmospheric CO₂. In light of this, we will discuss how the roles of CAs may have changed over time, focusing on the role of CA in pH regulation, how CAs affect CO₂ supply for photosynthesis and how CAs may help in the delivery of HCO₃⁻ for other metabolic reactions.

1. Introduction – the CA catalyzed reaction

Carbonic anhydrase (CA) is an enzyme that catalyzes the interconversion between CO₂ and HCO₃⁻ in solution as well as other reactions [1]. While the CO₂ and HCO₃⁻ conversion does take place in the absence of CA, the interconversion is very slow. It is thought that CA is required to ensure a rapid supply of CO₂ and HCO₃⁻ for various metabolic pathways in organisms.

2. There are multiple, apparently unrelated CA families

There are a surprisingly wide variety of CA proteins that fall into a number of protein families. These families are named by Greek letters and roughly follow the order in which they were discovered. The first class, the α-CA, was discovered in the 1930's in vertebrates. Work characterizing plant CAs in the early 1990s recognized this second group as a new type of CA [2,3]. A third group was identified in archaeobacteria in 1994 [4] and for a brief time the three protein families were referred to as the animal, plant and archaea CA forms. However, the discovery of both α- and β-CAs in *Chlamydomonas reinhardtii* and in terrestrial plants quickly showed that better terminology was needed and the Greek letters were quickly adopted [5]. The latest CA families, the η-CAs [6] and the θ-CAs [7,8], were discovered in 2015 and 2016,

respectively. It should be emphasized that each protein family appears to be phylogenetically unrelated to the others unless noted. In other words, when one compares the sequence or structure of the α- and γ-CA classes, there is often little to no sequence similarity or structural similarity [1]. Even the amino acids coordinating the Zn ions in the CA active sites are not conserved. However, in some cases the active site of some CAs are quite similar. For example, ζ and θ-CAs are structurally analogous to β-CAs at the active site while the active site of δ-CAs are similar to α-CAs.

2.1. Alpha CAs

The first CA was isolated from erythrocytes and later became known as a member of the α-CA class [9,10]. The α-CA is found in plants, green algae, diatoms, cyanobacteria and animals and is distinct from all other CA classes in both protein structure and amino acid sequence. The protein structure of α-CAs is dominated by a central β-sheet consisting of ten β-strands surrounded by seven peripheral α-helices [11]. The central β-sheet houses the active site of the α-CA, coordinating the zinc atom with three histidine residues and a water molecule [11]. Historically, the α-CA is regarded as the only CA class to not form multimers, but a few recent studies report the dimerization of α-CA monomers [12–14].

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2.2. Beta and epsilon CAs

The first report speculating the presence of CA in plants emerged in 1936 [15] but CA activity in plants was not observed in plants until 1939 when Arthur Neish successfully measured CA activity in isolated chloroplasts and whole leaf extracts of *Trifolium pretense*, *Arctium minus*, and *Onoclea sensibilis* [16]. The discovery that the prominent chloroplast CA is distinct from the α -class CA occurred much later via protein sequence analysis, and these CAs are now known as β -CAs [2,3,17]. The β -CAs are found in plants, algae, cyanobacteria, and non-photosynthetic bacteria but are excluded from animals. The β -CA monomer is distinct from α -CAs as it is mainly composed of α -helices that surround a central β -sheet comprised of four parallel β -strands [18]. β -CA monomers oligomerize to dimers to form two active sites consisting of a zinc atom coordinated by two cysteine residues, a histidine residue, and a water molecule [18]. These dimers can further interact to form tetramers and octamers [18,19]. ϵ -CAs are found in cyanobacteria and are now recognized as being a highly modified β -CA [20,21]. ϵ -CAs not only have an enzymatic function, they also form a structural part of the carboxysome shell in cyanobacteria [20].

2.3. Gamma CAs

γ -CAs were first discovered in the archaeon *Methanosarcina thermophila* [4]. Plants [22] and photosynthetic bacteria [23,24] also contain γ -CAs, whereas no reports have emerged detailing γ -CAs in animals. The archaeobacterial γ -CA possesses a zinc active site coordinated by three histidine residues and one water molecule [25]. To form the γ -CA active site, one monomer provides two zinc-coordinating histidine residues whereas a second monomer provides the third zinc-coordinating histidine residue [25]. Altogether, three monomers interact to form three active sites per homotrimer, the active γ -CA unit. The cyanobacterial protein CcmM is also a γ -CA, although modified. The N-terminus of CcmM has a γ -CA domain while the C-terminus has three to four RbcS domains allowing the protein to coordinate Rubisco packaging in the carboxysome. Espie and colleagues have shown that some CcmM proteins are enzymatically active CAs while others lack CA activity although they are still important in carboxysome packaging [26]. Those cyanobacteria with inactive CcmM proteins always have a β -CA in the carboxysome to convert HCO_3^- to CO_2 for Rubisco. In terrestrial plants, γ -CAs and γ -like CAs have been shown to be part of complex I of the mitochondria. Disruption of two γ -CA genes leads to loss of complex I. Green algae and diatoms also have γ -CAs localized to the mitochondria. However none of the CAs from eukaryotic algae or plants have been shown to have CA activity at this time.

2.4. Delta and zeta CAs

To date, δ -CAs and ζ -CAs have only been reported in diatoms and coccoliths [27,28]. The δ -CA, TWCA1 was reported in 1997 while the ζ -CA, CDCA1 was reported a few years later [29]. To date these CAs have not been found in other algae, even pennate diatoms. The Zn-binding region of CDCA1 is repeated three times and its structure was reported in 2012 [30]. An unusual aspect of the CDCA1 is its ability to use other metals besides Zn. Activity has been reported for this ζ -CA when binding others metals, most notably Cd. This ability to bind other metals might be an evolutionary adaptation to the low Zn levels often found in oceanic environments.

2.5. Eta CAs

Another new type of CA was reported in 2015. This CA, named an η -CA, was reported in the protozoan *Plasmodium falciparum* [31]. The structure of this η -CA has not been resolved yet, although the Zn coordination pattern is reported to be distantly related to that of α -CAs [32].

2.6. Theta CAs

θ -CAs are the most recent group of CAs reported. The θ class of CA has recently been described in the diatom *Phaeodactylum tricorutum* [7], the chlorophyte, *C. reinhardtii* [8] and the cyanobacterium *Chlorothece* [8]. In the diatom *P. tricorutum*, at least one of the θ -CAs, Pt43233, is localized to the thylakoid lumen [7]. Its function appears to be to catalyze the formation of CO_2 from the HCO_3^- pool in the diatom chloroplast. Since the pH of the thylakoid lumen is below the pKa of the CO_2 to HCO_3^- interconversion, this enzyme would tend to produce CO_2 from any HCO_3^- transported into the thylakoid. The loss of this protein results in a diatom growing slowly on air levels of CO_2 and showing a reduced affinity for inorganic carbon [7]. The other well-studied member of this CA family is the LCIB/LCIC complex of *C. reinhardtii* [8]. The *LCIB* gene (previously referred to as *Pmp*) encodes a θ -CA which is a chloroplast stromal protein surrounding the chloroplast pyrenoid [33]. Loss of *LCIB* results in a *C. reinhardtii* strain with a disabled CCM that requires high CO_2 concentrations for growth and photosynthesis [34]. *LCIB/LCIC* is thought to recapture CO_2 leaking out of the pyrenoid and possibly direct the HCO_3^- formed back into the pyrenoid [35].

3. CAs have a broad distribution in plants and algae

The CA enzyme is ubiquitous in nature although only certain classes of CA may be present in a particular organism. A distribution of the various CA classes is shown in Table 1. The focus of this table is to show whether a CA class is present in a group of plants or algae, but other selected model organisms are also shown for comparison. Since the sampling of many of the algal groups is limited to one or two genomes at this time, the absence of a specific class of CA in a specific lineage may not mean that all algae in that group lack that CA class. The origins of the α -, β - and γ -CA classes are ancient with all three CA types being found in bacteria, plants and many eukaryotic algal lineages. The ϵ -CA,

Table 1

Distribution of CAs in different lineages. One or more protein from each CA group was used to probe the genome sequences of each species using tBLASTn. Each species is only listed as having (+) or not having (–) a gene that encodes a protein related to that CA type. A species that is positive for a CA type may have multiple genes.

Taxa	Species	α CA	β CA	γ CA	δ CA	ζ CA	θ CA
Land plants	<i>Arabidopsis thaliana</i>	+	+	+	–	–	–
	<i>Oryza sativa</i>	+	+	+	–	–	–
Chlorophytes	<i>Chlamydomonas reinhardtii</i>	+	+	+	–	–	+
	<i>Coccomyxa subellipsoidea</i>	+	+	+	–	–	+
Prasinophytes	<i>Ostreococcus lucimarinus</i>	–	+	–	+	–	+
	<i>Micromonas pusilla</i>	–	+	+	–	+	+
Rhodophytes	<i>Chondrus crispus</i>	+	+	+	–	–	–
	<i>Galdieria sulphuraria</i>	+	–	+	–	–	–
Haptophyte	<i>Emiliania huxleyii</i>	+	–	+	+	–	+
Bacillariophytes	<i>Thalassiosira pseudonana</i>	+	–	+	+	+	+
	<i>Phaeodactylum tricorutum</i>	+	+	+	–	–	+
Phaeophyte	<i>Saccharina japonica</i>	+	+	–	–	–	–
Cryptomonad	<i>Guillardia theta</i>	+	+	+	–	–	+
Heterokont	<i>Aureococcus anophagefferens</i>	–	–	+	+	–	–
	<i>Escherichia coli</i>	–	+	+	–	–	–
Vertebrates	<i>Homo sapiens</i>	+	–	–	–	–	–
	<i>Mus musculus</i>	+	–	–	–	–	–
Insect	<i>Drosophila melanogaster</i>	+	–	–	–	–	–
	<i>Dictyostelium discoideum</i>	–	+	+	–	–	–
Fungi	<i>Saccharomyces cerevisiae</i>	–	+	–	–	–	–
	<i>Aphanomyces invadans</i>	+	+	+	–	–	–
Cyanobacteria	<i>Synechocystis</i> sp. PCC 6803	–	+	+	–	–	–
	<i>Cyanothece</i> sp. PCC8801	+	+	+	–	–	+
Archaea	<i>Sulfolobus acidocaldarius</i>	–	–	+	–	–	–
	DSM 639	–	–	+	–	–	–

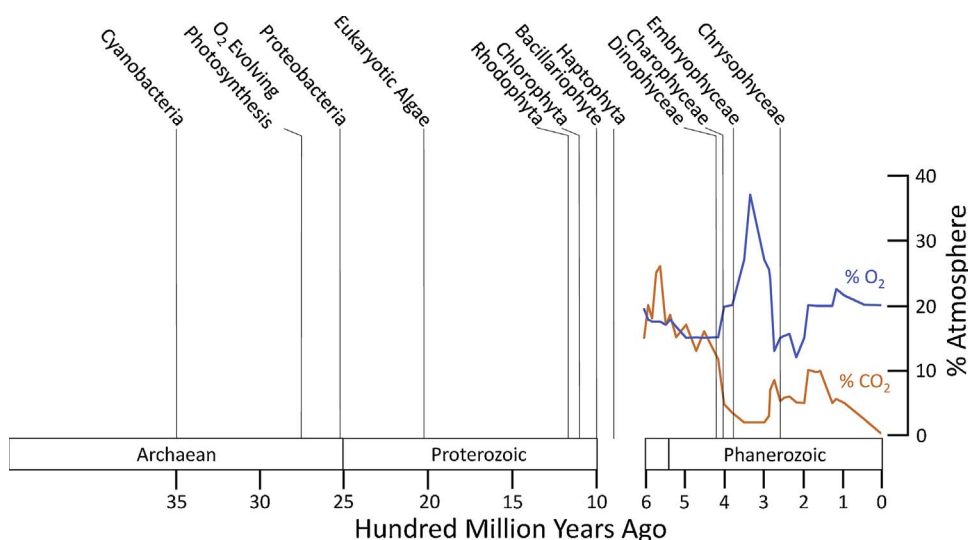


Fig. 1. The origins of different plant and algal lineages along with the historic atmospheric CO₂ and O₂ concentrations, based on Badger and Price [44] and Sage [45]. The CO₂ data are from Berner and Kothavala [43], the O₂ data from Berner [44] and the estimated ages of the algal lineages from Yoon et al. [50] and Bhattacharya et al. [51].

a variation on the β -CA class, is found in bacterial carboxysomes (Section 2.2). To date, the distribution of δ -CAs, ζ -CAs and η -CAs also appears to be more limited. It is interesting that genes encoding proteins with significant homology to the δ - and ζ -CAs can be seen in Prasinophytes, so these CAs are probably not limited to diatoms and coccoliths. In contrast, the newly discovered θ -CA class appears to be broadly distributed with paralogs exhibiting significant homology found in bacteria, cyanobacteria as well as eukaryotic algae (Table 1 and [8]).

It is important to note that the α -, β -, γ - and θ -CA classes are all found in organisms that diverged well over 1 billion years ago. The chlorophyte *C. reinhardtii* and the diatom *P. tricornutum*, diverged about 1.5 billion years ago, and both algal species have the α -, β -, γ - and θ -CA classes present. In addition, the α -, β -, γ - and θ -CA classes are all present in one or more types of bacteria, with cyanobacteria having representatives of all four CA classes. So these classes of CA are quite ancient although it is not clear why so many different CA families evolved in the first place.

The fact that there are so many classes of CAs raises the question why some organisms like vertebrates have only one type of CA (α) while photosynthetic organisms often have three or more CA families. There may be a number of reasons that many CA families have persisted in plants and algae. One reason is that a specific type of CA might be needed to interact with a larger complex. For example plant γ -CAs are part of complex 1 of the mitochondria and also play an intricate role, as CcmM, in the organization of Rubisco in the carboxysome in cyanobacteria. Another example would be CcaA, a β -CA also found in the carboxysome which has an unusual dimer of trimer structure and also interacts with CcmM [36]. It is possible that other CAs interact specifically with one or more enzymes thus ensuring that they are retained by the organism. Similarly, CAs are found in a variety of intracellular locations and in some cases bound to a specific cell membrane. For example, some of the CAs of plants and algae are found in the cell wall region. These are often α -CAs and perhaps they survive the environmental changes better than other CAs. Finally, some CAs are more active than others with α -CAs often having very high k_{cat} 's while β - and γ -CAs are less active. It is likely that many are under post-translational control as has been reported for CAH3 of *C. reinhardtii* [37].

4. Plants and many eukaryotic algae have a large number of CA isoforms

CAs have been studied in many plants and a limited number of eukaryotic algae. Plants, diatoms, and green algae all have many CA isoforms and all have α -, β -, and γ -CAs [38–41]. To date, 19 CA genes

have been identified in *Arabidopsis thaliana* (8 alpha, 6 beta, and 5 gamma genes) [38,40], 19 CA genes have been identified in *C. reinhardtii* (3 alpha, 6 beta, 4 gamma, and 4 theta-CAs) [41], and the diatom *P. tricornutum* has 13 genes (5 alpha, 2 beta, 2 gamma, and 4 theta-CAs) [42]. Notably, these CA genes encode proteins that localize to many compartments within the cells of these eukaryotic organisms. *Arabidopsis* has CA isoforms in the plasma membrane, cytoplasm, mitochondria, and chloroplast stroma [38,40]. *Chlamydomonas* has isoforms in the cell wall, plasma membrane, cytosol, mitochondria, chloroplast stroma, and chloroplast thylakoid lumen [41]. Most of this work has been done using CA-GFP fusions and it has not been determined whether the CA active site is on the cytoplasmic or exoplasmic side of the membrane. *P. tricornutum* has CA isoforms in most of the same locations. In addition, some of the *P. tricornutum* CAs are in intermembrane spaces within the chloroplast as diatom chloroplasts are surrounded by four sets of membranes instead of two [7,42]. While the CA isoforms of other algae have not been localized, a check of the available genomes indicates they also have a similarly large number of CA genes and these genes likely encode CAs that go to many locations within the cell.

5. CAs may have taken on new roles when the atmospheric CO₂ concentration decreased

The Earth's atmospheric CO₂ level decreased significantly starting around 400 million years ago and then decreased to new low levels during the past 50 million years (Fig. 1) [43–46]. This decrease in CO₂ occurred well after most of the algal lineages listed in Table 1 diverged [47–51]. Therefore, terrestrial plants as well as the different types of algae had to independently adapt to the changes in the Earth's atmospheric composition. Evidence for this independent adaptation comes from the different types of CCMs and how the various CAs are used in the CCMs of different algae. For example, the α -CAs CAH3 [52] and CAH1 [53], appear to be responsible for the dehydration of HCO₃⁻ inside the lumen of the thylakoid in *C. reinhardtii* and *Nannochloropsis oceanica*, respectively. In contrast, the θ -CA Pt43233 is localized to the thylakoid lumen of *P. tricornutum* [7] and is carrying out the same role as the α -CAs CAH3 and CAH1, namely supplying CO₂ from accumulated HCO₃⁻ thereby increasing the CO₂ supply for Rubisco.

This is similar to what is seen with C₄ plants, although perhaps on a longer timescale. There is good evidence that C₄ photosynthesis has evolved independently at least 60 times in the past 40 million years [45,54]. Algal CCMs may have evolved even earlier as algae have dealt with perhaps two low CO₂ events over the past 400 million years (Fig. 1). In addition, since CO₂ diffuses 10,000 times slower in water

Table 2

There are multiple enzymes in plant cells that use either CO₂ or HCO₃[−] to carboxylate their substrate. The measured K_m (CO₂ or HCO₃[−]) for the enzymes is given as values in the mM range.

Pathway	Enzyme	Form of C _i Used	Cofactor	K _m (mM)	References
Photosynthesis	Rubisco	CO ₂	None	0.009–0.120	[66,67]
Photosynthesis (C ₄)/Amino acid metabolism	Phosphoenolpyruvate Carboxylase (C ₄)	HCO ₃ [−]	Mg ²⁺	0.027–0.180	[68–72]
Fatty Acid Synthesis	Acetyl-CoA Carboxylase	HCO ₃ [−]	Biotin	0.9–2.5	[73–75]
Leucine Catabolism	3-methyl Crotonyl-CoA Carboxylase	HCO ₃ [−]	Biotin	0.8–2.0	[76]
Unknown	Geranoyl-CoA Carboxylase	HCO ₃ [−]	Biotin	0.6	[76]
Urea Catabolism	Urea Amidolyase	HCO ₃ [−]	Biotin	2.0	[77]
Pyrimidine and Arginine Synthesis	Carbamoyl Phosphate Synthase	HCO ₃ [−]	None	1.7	[78]
Purine Synthesis	5-aminoimidazole Ribonucleotide Carboxylase	HCO ₃ [−]	None	23.0	[79]

than air, algae have experienced a stronger selective pressure towards the use of a CCM. In light of this, perhaps it is not surprising to see different CA proteins taking on similar metabolic functions in different algal lineages as the CO₂ concentration dropped. In addition, as the atmospheric CO₂ concentration dropped, the supply of both CO₂ and HCO₃[−] decreased. The result is obvious in algae as the vast majority of aquatic photosynthetic organisms possess a CCM and all of these CCMs rely on one or more CAs. However, the effect of the decreases in atmospheric CO₂ has had on HCO₃[−] is less well understood. As discussed in the next section, many enzymes utilizing HCO₃[−] have relatively poor affinity for HCO₃[−] and the supply of HCO₃[−] for these reactions might be problematic in the absence of CA activity. Finally, the current low CO₂ concentration raises the question of whether the CO₂ and HCO₃[−] buffering system is actually important in plants.

5.1. CAs play essential roles in the CO₂ concentrating mechanism of algae and plants

One of the best understood roles of CA in cyanobacteria and algae is in the delivery of CO₂ to Rubisco. In cyanobacteria, a CA is always localized to the carboxysome, where it converts HCO₃[−] entering the carboxysome to CO₂ for use by Rubisco. In some cyanobacteria like *Synechococcus* PCC7942, this CA is a β-CA [55], in others such as *Prochlorococcus*, it is a ε-CA [20], while in the cyanobacteria *Nostoc*, the CA is a modified γ-CA [26]. In eukaryotic algae, the CA responsible for converting accumulated HCO₃[−] to CO₂ for Rubisco is found in the chloroplast. This is accomplished by an α-CA, CAH3, in *C. reinhardtii* [52]. *Nannochloropsis oceanica* also has an α-CA, CAH1 that supplies CO₂ for Rubisco from accumulated HCO₃[−]. However, in *P. tricornutum*, the θ-CA, Pt43233, appears to perform the same function [7]. These proteins are all localized in the thylakoid lumen. Another protein complex, LCIB/C also plays a key role in the CCM of *C. reinhardtii*. This θ-CA-like protein is found in the stroma and has a close association with the chloroplast pyrenoid. It is thought that it may recapture CO₂ leaking out of the pyrenoid [35].

Another role of CA in the CCMs of cyanobacteria and algae is to facilitate the entry of CO₂ from the environment. These CAs are found in the cell walls or periplasmic spaces of the organism and convert HCO₃[−] to CO₂ increasing the CO₂ availability to the alga. Again a variety of CAs are found in the walls of different algae. In *C. reinhardtii*, α-CAs are found in the cell wall, helping deliver C_i across the plasma membrane [41,56–58]. In some species of diatoms, β-CAs are found in the cell walls, while in other species the cell wall CAs are of the δ and/ or ζ variety [59,60].

In terrestrial C₄ plants, a β-type CA in the cytosol of leaf mesophyll cells converts incoming CO₂ to HCO₃[−] to be used by PEP carboxylase [61]. The role of CA in terrestrial C₃ plants is less understood seemingly due to the compensatory effect of multiple CA isoforms. Studies of single knockout CA lines of C₃ plants do not show apparent phenotypic

responses under normal growth conditions [62–65], whereas removing multiple CAs have generated a measurable response to various environmental conditions [62–65]. Considering the large number of CA genes potentially coding for active CA proteins, it is possible that removing a single CA isoform does not reduce CA activity nor produce an observable phenotypic response because another CA may compensate for the loss of the other CA. The fact that multiple CA genes may be expressed in a certain cell type [62] or multiple CA isoforms may be present in the same subcellular location [44,64] calls for additional studies looking at multi-knockout CA lines to characterize CA functionality in plants.

5.2. As atmospheric CO₂ dropped, HCO₃[−] became a limiting substrate

When we think of the role of CA in plants and algae we tend to think of photosynthesis and to a lesser extent respiration and photorespiration. To be sure, these metabolic pathways are the largest consumers and producers of CO₂ in the plant cell. However, other pathways also require C_i, usually in the form of HCO₃[−]. Some of these pathways and the enzymes that use HCO₃[−] are shown in Table 2.

There are seven enzymes in Table 2 that utilize HCO₃[−] as a source of C_i in carboxylation reactions. All of the enzymes that have HCO₃[−] as a substrate have been found in a variety of plants, except urea amidolyase, which so far has only been found in green algae [80]. While PEP carboxylase is found in the cytoplasm of plants, most of the enzymes listed in Table 2 have been shown to be in plastids or are thought to be in plastids as they have a chloroplast leader sequence. Four of the six enzymes that use HCO₃[−] are biotin-dependent carboxylases [76]. The first step in biotin-dependent carboxylation reactions is the carboxylation of biotin. The second step in the reaction is the transfer of the carboxyl group from biotin to an acceptor molecule which gives rise to the name of the enzyme. For example, for acetyl-CoA carboxylase the acceptor molecule is acetyl-CoA. In all biotin-dependent carboxylases the protein entity (domain or subunit) that catalyzes the carboxylation of biotin has a three-dimensional structure that places it in the ATP grasp superfamily of enzymes [81]. Enzymes in the ATP grasp superfamily catalyze reactions involved in coupling of a carboxylate to an amino (or thiol) group. The ATP is used to activate the carboxylate for nucleophilic attack by formation of an acylphosphate intermediate (carboxyphosphate where HCO₃[−] is the carboxyl group) [81]. The nucleophile for the biotin-dependent carboxylases is the N1' of biotin. The molecular architecture of the ATP grasp superfamily of enzymes is also used for two HCO₃[−] dependent carboxylases that do not use biotin as the nucleophile. Carbamoyl phosphate synthetase and 5-aminoimidazole ribonucleotide carboxylase use ammonia and a primary amine as the nucleophiles, respectively [82].

A question that arises after compiling a list of these enzymes is whether CA is required to supply HCO₃[−] in high enough concentrations to efficiently run these reactions. Would the HCO₃[−] concentration be

high enough in the absence of CA activity to maintain growth and viability of the cell? Photosynthetic organisms have faced very low atmospheric CO₂ concentrations for long periods of time over the past 600 million years (Fig. 1). At 200 ppm the concentration of CO₂ dissolved in water at 25 °C is approximately 6 μM. If this solution is buffered at pH 7.2–7.3, typical of the cytoplasm, the HCO₃[−] concentration at equilibrium would be 60 μM. At the current atmospheric CO₂ concentration those numbers would be about double, 12 μM for CO₂ and 120 μM for HCO₃[−]. Nevertheless, if one compares the expected historical concentrations of CO₂ and HCO₃[−] to the K_m(HCO₃[−]) for the various carboxylating enzymes (Table 2), it would appear that the carboxylating enzymes were often operating below their K_m(HCO₃[−]) or K_m(CO₂). In the absence of CA activity, the HCO₃[−] supply for the reactions may be even lower. In fact, even in the presence of CA those enzymes that utilize HCO₃[−] as a source of CO₂ would only be about 0.5 – 17% saturated at 120 μM HCO₃[−] and, as a result, most of the catalytic power of the carboxylating enzymes (Table 2) would be wasted. Again, the situation would be worse in the absence of CA. It is important to note that the low degree of saturation with HCO₃[−] is particularly acute for the two enzymes involved in nucleotide synthesis, an essential pathway for all organisms. At a concentration of 120 μM HCO₃[−], 5-aminoimidazole ribonucleotide carboxylase and carbamoyl phosphate synthetase would only be 0.5 and 6.6% saturated with HCO₃[−], respectively. Therefore, CA could play a significant role in supplying the HCO₃[−] needed for these reactions, resulting in increased enzymatic activity and ultimately more robust cellular growth.

In the case of Rubisco, which is clearly operating at very poor efficiency, plants and algae have evolved ways to increase the CO₂ concentration to maintain photosynthesis. The fact that most aquatic photosynthetic organisms have a CCM and that C4 photosynthesis has independently arisen over 60 times strongly supports the idea that the low atmospheric CO₂ level put plants and algae under enormous selective pressure. What about the other carboxylating enzymes in plants and algae? Part of the answer to that question might be found in the study of yeast and *Escherichia coli* CA knockout strains.

Yeast has one known gene NCE103, encoding a CA. This CA has 60% sequence similarity to the β-type CAs in plants [83,84]. While the gene is nuclear encoded, the protein was first localized to the cytosol [85] where it has been reported to co-localize with several carboxylating enzymes. However, a recent proteomic study localized this protein to the intermembrane space (IMS) of the mitochondria [86]. Analysis of the N-terminal sequence of the protein using a prediction program (YLOC) reveals no mitochondrial target sequence, which is consistent with a targeting to the IMS or to the cytosol. If the single CA gene in yeast is mutated, the mutant (*Δnce103*) cannot grow on air levels of CO₂, a phenotype rescued by growing the cultures on elevated CO₂ [83,84,87]. Further studies with this *Δnce103* mutant confirmed that the basis for this phenotype was that the CA provided HCO₃[−] for various carboxylation reactions such as those catalyzed by pyruvate carboxylase, acetyl-CoA carboxylase, and carbamoyl-phosphate synthase [84,88]. Some of these reactions are shown in Table 2. Since CO₂ can readily cross cell membranes, it appears that the CO₂ concentration in the cytosol remains fairly low when the external CO₂ concentration is at ambient levels or below, and the CO₂ generated in metabolic reactions is not sufficient to support growth.

E. coli also require a continuous supply of inorganic carbon for growth, and, like yeast, the source of HCO₃[−] needed for these reactions is from CO₂ hydration catalyzed by CAs. Two genes, *Can* and *CynT* encode β-type CAs in *E. coli* [89,90]. Single knockout mutants of these two genes have provided insight into the functions of the CAs in *E. coli*. The *Can* gene is constitutively expressed and mutant cells cannot grow normally on air levels of CO₂. In contrast, the *CynT* gene is induced only during cyanate metabolism, and the mutant requires high CO₂ to grow in the presence of cyanate [90]. The growth of both mutants is poor due to inhibition of metabolic functions involving HCO₃[−] [91], implying the CAs function to recapture the released CO₂, and converting it to

HCO₃[−] [92]. A high reversion rate has however, been reported for the *can* mutant due to expression of *CynT* [90]. As expected, if the two genes are silenced, the double mutant cannot grow on ambient CO₂. Reintroduced CAs have been able to increase the flux of HCO₃[−] in *E. coli*, restoring growth on air [92]. In a complementation assay for HCO₃[−] uptake, cyanobacteria transporters were also able to rescue the growth phenotype of a high CO₂ requiring mutant in *E. coli* [93], confirming the main role of the CAs is for HCO₃[−] supply. That yeast and *E. coli* CA knockout strains must be grown on elevated CO₂ concentrations to survive is almost counter-intuitive considering that they both are producing CO₂ from respiration. That coupled with the loss of CA means the production of HCO₃[−] is greatly reduced and it appears likely that it is the low HCO₃[−] concentration that leads to the reduced growth phenotype on air.

5.3. Do CAs help control the internal pH in cells at low atmospheric CO₂?

One function often assigned to CAs is regulation of pH. Is this the case in plant cells? To some extent, plant scientists have been overly influenced by medical researchers and vertebrate physiologists regarding the importance of CO₂ and HCO₃[−] in controlling pH. To be sure, CA plays an important role in buffering blood pH and probably the cytoplasm and intracellular compartments for most animals. The CO₂ partial pressure (4.7–6.0 kPa) in the blood is typically over 100 times the atmospheric CO₂ partial pressure (40 Pa). At pH 7.4, the typical pH of the blood, the HCO₃[−] concentration averages about 24 mM while the dissolved CO₂ concentration is close to 1.5 mM. Thus, in the blood, CO₂ and HCO₃[−] are crucial to buffering the system. Likewise, in the oceans, CO₂ and HCO₃[−] also play a major role in determining the pH of the seas. Of course, the CO₂ concentration of the ocean is much lower than blood by a factor of about 100, but since the pH of seawater is almost one unit higher than blood, the HCO₃[−] concentration remains in the millimolar range. In addition, there are few other ions in seawater with pK_as close to 8.2 that are present in high enough concentrations to strongly buffer the ocean.

What about plant cells and freshwater algae? In those organisms the CO₂ concentration is very low, about 12 μM. At the typical pH of the cytoplasm, the HCO₃[−] concentration would only be about 110 μM. Even in the chloroplast stroma, the HCO₃[−] concentration would only be a bit over 1 mM at equilibrium. Thus, the buffering capacity of CO₂ and HCO₃[−] is only about 1/100 of the buffering capacity of the same system in the blood. Since many metabolites of the cytoplasm and the other organelles are in the 1–10 mM range and often phosphorylated, it would appear that CO₂ and HCO₃[−] are unlikely to be the major buffering metabolites in plant cells or freshwater algae. A possible exception might be fresh-water cyanobacteria which can concentrate HCO₃[−] in the cytoplasm a thousand fold. However, they don't have a CA in the cytoplasm which would appear to reduce the role of HCO₃[−]/CO₂ as a cytoplasmic buffer.

6. Conclusions

Progress has been made in elucidating physiological roles for CA isoforms in green algae, photosynthetic bacteria, and C₄ plants, all organisms which contain CCMs. Characterizing roles for CA in C₃ plants which lack a recognizable CCM has proven to be more difficult, albeit researchers have reported some non-photosynthetic roles for CAs in C₃ plants. Hopefully increased utilization of T-DNA libraries for Arabidopsis and new techniques such as the CRISPR- Cas-9 system can yield a new wave of CA research in C₃ plants.

A recent topic of considerable interest is the introduction of an algal or cyanobacterial CCM in higher plants to increase their photosynthetic rates. Researchers are progressing with these projects [94,95] although there is still a ways to go before an algal- or cyanobacterial-like CCM can be realized in a C₃ plant. Native CA activity in C₃ chloroplasts has the potential to disrupt the functionality of an imported algal or

cyanobacterial CCM in the same way as overexpressing human CA in the cytosol impacted the cyanobacterial CCM [96]. The pertinent solution is to remove native C_3 chloroplast CA activity to prevent futile cycling of the different C_i species [97] as most studies that reduced C_3 chloroplast CA activity displayed no negative effects in the plants. Since CAs are referred to as some of the fastest enzymes known, reducing CA activity by 90% may still yield a high enough CA activity to maintain physiological processes. Since CA activity in C_3 plants is not well understood, a greater understanding of the roles of CA in different metabolic reactions is needed before any CA isoform is removed for the implementation of a CCM. With the many CA isoforms in C_3 plants, there is great opportunity for the discovery of new roles for CA.

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